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Standards

Irish Standard Recommendation  
S.R. CEN/TR 17225:2018

Fuels and biofuels - Assessment on oxidation stability determination methods for distillate fuels and blends thereof with fatty acid methyl esters (FAME)

**S.R. CEN/TR 17225:2018**

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## National Foreword

S.R. CEN/TR 17225:2018 is the adopted Irish version of the European Document CEN/TR 17225:2018, Fuels and biofuels - Assessment on oxidation stability determination methods for distillate fuels and blends thereof with fatty acid methyl esters (FAME)

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TECHNICAL REPORT

CEN/TR 17225

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## Fuels and biofuels - Assessment on oxidation stability determination methods for distillate fuels and blends thereof with fatty acid methyl esters (FAME)

Carburants et biocarburant - Assessment des méthodes de déterminations de la stabilité à l'oxydation pour distillats et mélanges avec esters méthyliques d'acides gras (EMAG)

Mitteldestillatkraft- und -brennstoffe und Biokraftstoffe - Bewertung der Verfahren zur Oxidationsstabilitätsbestimmung für Mitteldestillatkraft- und -brennstoffe und deren Mischungen mit Fettsäure-Methylestern (FAME)

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<b>Contents</b>	<b>Page</b>
<b>European foreword</b> .....	<b>3</b>
<b>Introduction</b> .....	<b>4</b>
<b>1 Scope</b> .....	<b>9</b>
<b>2 Normative references</b> .....	<b>9</b>
<b>3 Terms and definitions</b> .....	<b>9</b>
<b>4 Test methods quantifying fuel ageing products</b> .....	<b>10</b>
<b>4.1 History</b> .....	<b>10</b>
<b>4.2 Generic oxidation stability test</b> .....	<b>10</b>
<b>4.3 Turbine and mineral oil test</b> .....	<b>11</b>
<b>4.4 Inherent storage stability</b> .....	<b>12</b>
<b>4.5 Delta acid method and acid number determination after ageing at 115 °C</b> .....	<b>12</b>
<b>4.6 High temperature stability</b> .....	<b>13</b>
<b>5 Test methods determining the ageing reserve on the fuel</b> .....	<b>13</b>
<b>5.1 General</b> .....	<b>13</b>
<b>5.2 Rancimat</b> .....	<b>13</b>
<b>5.3 Rancimat at 120 °C</b> .....	<b>14</b>
<b>5.4 PetroOxy</b> .....	<b>14</b>
<b>6 Differences in Rancimat and PetroOxy tests</b> .....	<b>14</b>
<b>6.1 General</b> .....	<b>14</b>
<b>6.2 Experimental studies on the share of fuel ageing at the Induction Period</b> .....	<b>16</b>
<b>6.3 Discussion of pressure curve characteristics for the PetroOxy-Test (IPPetroOxy)</b> .....	<b>17</b>
<b>6.4 Discussion of conductivity curve characteristics for the Rancimat test (IPRancimat)</b> .....	<b>18</b>
<b>7 Correlation between Rancimat (EN 15751) and PetroOxy (EN 16091) test methods</b> .....	<b>21</b>
<b>8 Synopsis</b> .....	<b>23</b>
<b>Annex A (informative) Share of fuel ageing on the Induction Period of the Rancimat- and PetroOxy-tests</b> .....	<b>25</b>
<b>Annex B (informative) Correlation exercise between PetroOxy and Rancimat</b> .....	<b>29</b>
<b>Bibliography</b> .....	<b>31</b>

## **European foreword**

This document (CEN/TR 17225:2018) has been prepared by Technical Committee CEN/TC 19 “Gaseous and liquid fuels, lubricants and related products of petroleum, synthetic and biological origin”, the secretariat of which is held by NEN.

Attention is drawn to the possibility that some of the elements of this document may be the subject of patent rights. CEN shall not be held responsible for identifying any or all such patent rights.

During revisions of standards EN 14112, EN 15751, EN 16568 (Rancimat, [17]) and EN 16091 (PetroOxy), CEN/TC 19/JWG 1 “Vegetable fats and oils and their by-products for use in automotive fuels (Joint working group with CEN/TC 307)” felt the need to create a document that provides extended background information on oxidation stability test methods complementing the abstracts given in Annex A of EN 14112, EN 15751 and EN 16568 as well as in Annex C of EN 16091.

This report also gives an overview about the work done by CEN/TC 19/JWG 1 on the correlation between the Induction Period of the Rancimat and PetroOxy test methods.

## CEN/TR 17225:2018 (E)

### Introduction

All fuels, whether of fossil source or biogenic origin, constantly degrade. Although the bulk composition remains widely unchanged, the presence of oxygen, higher temperature and catalytically active metals accelerate oxidation of less stable compounds present, thus generating ageing products like acids oligomers and polymeric compounds. The complex and varying composition of fossil diesel fuels and the interactions of FAME and its inherent side-products with base diesel fuel components make it difficult to understand fuel ageing mechanisms at a molecular level. The determination of oxidation stability of fuels is therefore addressed by phenomenological techniques. Most test methods to characterize fuel oxidation stability are based on the determination of specific parameters, e.g. polymer formation, acidity increase, or oxygen consumption.

It has been established that the phenomenon of fuel ageing consists of two consecutive phases, starting with the depletion of the ageing reserve with few chemical changes to the bulk material, followed by the fuel ageing process itself during which the fuel is badly decomposed forming ageing polymers and acids (Figure 1, Table 1). Fuel oxidation takes place via a free radical chain process, initiated by the abstraction of a hydrogen atom from the fuel molecule and the addition of molecular oxygen to form hydroperoxides.

Following chemical reactions proceed into two directions: a) Fragmentation of the molecule generates aldehydes, ketones and short-chain carboxylic acids; b) interlinking of the molecules yielding oxygen bridged dimers and polymers.

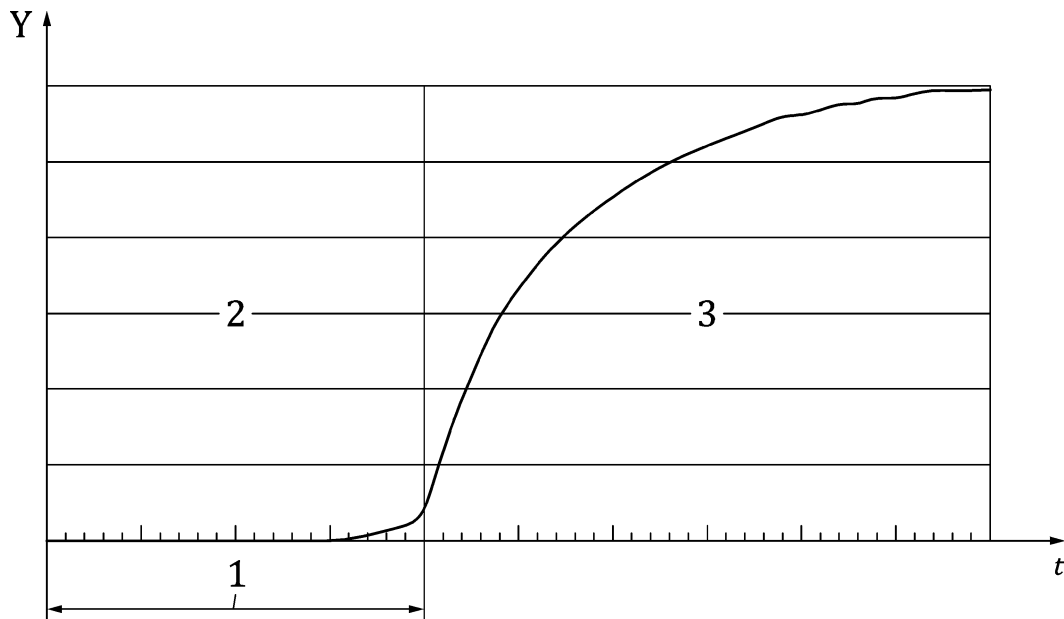
Diesel fuel oxidation is influenced by the chemical constituents of diesel fuel (e.g. specific sulphur- and nitrogen-containing compounds) and contaminants (e.g. metal impurities). There are many different chemical reactions, which can occur simultaneously and sequentially, fuel molecules are either fragmented or increased in their size, e.g. by di-, oligo- and polymerization. These reactions rapidly increase after a period when little or no chemical oxidation took place. This time interval before fuel oxidation gets started significantly is called Induction Period.

Stress imposed on the fuel by heat in the presence of oxygen does not necessarily result in immediate fragmentation or dimer/polymer formation. The reserve capacity of the fuel to resist oxidation is called ageing reserve, which is related to the Induction Period [1, 2].

Test methods in Figure 1 marked with an asterisk are mentioned in this report for additional information, but are not the focus of this work.



Rancimat EN 14112 EN 15751 EN 16568	EN ISO 12205 [18], ASTM D2274 [19] IP 306 mod.* [20]
PetroOxy EN 16091 ASTM D7545 [21]	$\Delta$ -acid no. method, JIS draft* [8], CEN/TR 16885 [10], ASTM D6468* [22]



**Key**

- 1 period for the depletion of the ageing reserve (Induction Period)
- 2 ageing reserve
- 3 fuel ageing process, strong formation of acids and polymers
- t time, h
- Y property (such as acid number, conductivity, acidity, polymer formation, etc.)

**Figure 1 — Two consecutive phases of fuel ageing and test methods relevant for each phase**

## CEN/TR 17225:2018 (E)

Table 1 — Overview about parameters and conditions of oxidation stability test methods

	<b>Rancimat EN 14112 EN 15751 EN 16568</b>	<b>PetroOxy EN 16091 ASTM D7545</b>	<b>EN ISO 12205 ASTM D2274</b>	<b>IP 306</b>	<b>IP 306 mod.</b>	<b>ASTM D4625</b>	<b><math>\Delta</math>-Acid method JIS draft</b>	<b>ASTM D6468</b>
Title	Determination of oxidation stability by accelerated or rapidly accelerated oxidation method	Determination of oxidation stability by rapid small scale oxidation test (RSSOT)	Determination of the oxidation stability of middle distillate fuels	Determination of oxidation stability of straight mineral oil	Modifications proposed in [3] to adapt IP 306 to diesel fuels, FAME blends and neat FAME	Standard test method for distillate fuel storage stability at 43 °C	Determination of oxidation stability of FAME-blended diesel fuels	Standard test method for high temperature stability of distillate fuels
Scope/Determination	Oxidation stability by means of measuring the Induction Period	Stability under accelerated oxidation conditions by measuring the Induction Period	Inherent stability of middle distillate petroleum fuels under accelerated oxidation conditions	Tendency of straight (i.e. plain) mineral oil to oxidize, expressed as total oxidation products (TOP)	Inherent stability under accelerated oxidation conditions	Inherent storage stability of distillate fuels under mild aging conditions with limited air exposure	Acid value increase after accelerated oxidation conditions	Relative stability of middle distillate fuels under high temperature aging conditions with limited air exposure (venting)
FAME content <sup>a</sup>	B100 (EN 14112)  B2 to B100 (EN 15751)  B2 to B50 (EN 16568)	B0 to B100	B0 (EN ISO 12205 ASTM D2274)  B0 to B7	B0	B0 to B100 [3]	B0	B0 to B5	B0

	<b>Rancimat</b> <b>EN 14112</b> <b>EN 15751</b> <b>EN 16568</b>	<b>PetroOxy</b> <b>EN 16091</b> <b>ASTM D7545</b>	<b>EN ISO 12205</b> <b>ASTM D2274</b>	<b>IP 306</b>	<b>IP 306 mod.</b>	<b>ASTM D4625</b>	<b>Δ-Acid</b> <b>method JIS</b> <b>draft</b>	<b>ASTM D6468</b>
Oxidation stability parameter	Ageing reserve	Ageing reserve + partial fuel ageing	Fuel ageing behavior by fuel oxidation products	Fuel ageing behavior by fuel oxidation products	Fuel ageing behavior by fuel oxidation products	Fuel ageing behavior by fuel oxidation products	Fuel ageing behavior by fuel oxidation products	Fuel ageing behavior by fuel oxidation products
Parameter	Induction Period by conductivity increase in water cell	Induction Period by oxygen pressure drop	Amount of total insoluble sludge	Amount of total insoluble sludge + volatile acidity + soluble acidity	Amount of total insoluble sludge + sludge from soluble polymers + volatile acidity + soluble acidity	Amount of total insoluble sludge	Acid increase by soluble acidity	light reflectance by filterable insoluble sludge
Sample amount	3,0 g (EN 14112) 7,5 g (EN 15751) 7,5 g (EN 16568)	5 ml	350 ml	25 g	25 g	400 ml	350 ml	50 ml
Temperature	110 °C (EN 14112) 110 °C (EN 15751) 120 °C (EN 16568)	140 °C	95 °C	120 °C	120 °C	43 °C	115 °C	150 °C
Time	—	—	16 h	48 h	16 h	up to 24 weeks	16 h	1.5 or 3 h

## CEN/TR 17225:2018 (E)

	<b>Rancimat</b> <b>EN 14112</b> <b>EN 15751</b> <b>EN 16568</b>	<b>PetroOxy</b> <b>EN 16091</b> <b>ASTM D7545</b>	<b>EN ISO 12205</b> <b>ASTM D2274</b>	<b>IP 306</b>	<b>IP 306 mod.</b>	<b>ASTM D4625</b>	<b>Δ-Acid</b> <b>method JIS</b> <b>draft</b>	<b>ASTM D6468</b>
Oxidant	10 l/h Air	700 kPa Oxygen	3 l/h Oxygen	1 l/h Oxygen	1 l/h Air	ambient air pressure, no blow-through	3 l/h Oxygen	ambient air
Catalyst	no	no	no	no	no	no	no	no
Dilution of aged fuel	—	—	no	no	10 to 20-fold excess of heptane for Bx	no	—	—
Precision (reproducibility) <sup>b</sup>	0,26 X + 0,23 (EN 14112) 0,1903 8 X + 0,3726 9 (EN 15751) 0,139 5 X + 0,288 8 (EN 16568)	0,086 3 X + 1,377 2 (EN 16091)	10,6 (0,1 X) <sup>0,25</sup> EN ISO 12205	precision for insoluble sludge not reported	no full validation performed	2,20√X	no precision available	-0,428 1 X + 44,08 (1,5 h) -0,303 4 X + 34,11 (3,0 h)

Total insoluble sludge is defined as solid sediment (sum of filterable sludge and sludge adherent to the inner walls of the glass tube).

Sludge from soluble polymers is defined as the amount of polymers dissolved in the aged fuel but precipitated as sludge after dilution of the aged fuel with heptane.

Soluble acidity is defined as the acidity in the aged fuel sample.

Volatile acidity is defined as the acidity in the collecting vessel with the water phase.

<sup>a</sup> FAME content which is validated with precision statement. Method can be applicable for blends containing higher FAME contents.

<sup>b</sup> The precision of the test methods is given as complimentary information. Parameters and conditions of oxidation stability test methods are different limiting the significance of a direct comparison of precision statements. x is the mean of two results.

## 1 Scope

This document provides an overview of existing oxidation stability methods, with an emphasis on differences between the Rancimat (EN 14112/EN 15751) and PetroOxy (EN 16091) tests.

## 2 Normative references

The following documents are referred to in the text in such a way that some or all of their content constitutes requirements of this document. For dated references, only the edition cited applies. For undated references, the latest edition of the referenced document (including any amendments) applies.

EN 14112, *Fat and oil derivatives - Fatty Acid Methyl Esters (FAME) - Determination of oxidation stability (accelerated oxidation test)*

EN 15751, *Automotive fuels - Fatty acid methyl ester (FAME) fuel and blends with diesel fuel - Determination of oxidation stability by accelerated oxidation method*

EN 16091, *Liquid petroleum products - Middle distillates and fatty acid methyl ester (FAME) fuels and blends - Determination of oxidation stability by rapid small scale oxidation method*

## 3 Terms and definitions

For the purposes of this document, the following terms and definitions apply.

ISO and IEC maintain terminological databases for use in standardization at the following addresses:

- IEC Electropedia: available at <http://www.electropedia.org/>
- ISO Online browsing platform: available at <http://www.iso.org/obp>

### 3.1

#### **ageing reserve**

reserve capacity of the fuel to resist oxidation

### 3.2

#### **break point**

in the Rancimat test, the break point reflects the moment of beginning oxidation when the formation of volatile acids increases rapidly; and in the PetroOxy test, the break point is the pressure which is 10 % below the maximum pressure observed in the test apparatus

Note 1 to entry: The break point is differently defined depending on the testing procedure.

### 3.3

#### **fuel ageing**

oxidation of the fuel, forming ageing acids and polymers

### 3.4

#### **Induction Period**

IP

time elapsed until significant oxidation of fuel begins

### 3.5

#### **inter-laboratory study**

ILS

ring tests in which independent laboratories gather precision data for test methods using the same set of subsamples

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